**Stoichiometry: Limiting and Excess: Quiz 5a**

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**Make sure to SHOW ALL WORK, INCLUDE UNITS and correct SIG FIGS!**

*For the following please include a BALANCED EQUATION*

1. \_\_\_\_\_ CaO + \_\_\_\_\_ H2O 🡪 \_\_\_\_\_ Ca(OH)2

Identify the limiting and excess reactants when 4.44 g of calcium oxide and 7.77 g of water are available to react. **How many grams of calcium hydroxide will be formed in this reaction?** *Make sure to state which of the reactants is limiting and which is in excess.*

Answers:



