**Stoichiometry: Mole to Mole Conversions: Quiz 2a**

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**Make sure to SHOW ALL WORK, INCLUDE UNITS and correct SIG FIGS!**

*For the following please include a BALANCED EQUATION*

1. A student places 7.50 moles of chromium (III) metal in an excess of concentrated hydrochloric acid. How many moles of hydrogen gas are formed?

Answers:



1. 