**Unit 6: Solution Chemistry Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Blk: \_\_\_\_\_\_**

| **Topic** | **Checkpoints** | | | **Plan for Improvement** |
| --- | --- | --- | --- | --- |
| **1) Polarity – Intermolecular Forces** |  |  |  |  |
| **2) VSEPR – Shapes of Molecules (BEG)** |  |  |  |  |
| **3) VSEPR – Shapes of Molecules (ADV)** |  |  |  |  |
| **4) Dilutions** |  |  |  |  |
| **5) Titrations** |  |  |  |  |

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| --- | --- | --- | --- | --- | --- |
| **TEACHER: Please do not write in these rows!** | ***Learner was unable to reflect in any ways described to the right*** | **Novice** | **Learner** | **Practitioner** | **Expert** |
| Learner can rarely evaluate their own strengths & weaknesses and does not have a clear view on how to improve their learning | Learner can somewhat evaluate their own strengths & weaknesses and has a somewhat clear view on how to improve their learning | Learner can evaluate their own strengths & weaknesses and has specific ideas on how to improve their learning | Learner can evaluate their own strengths & weaknesses consistently and independently and has specific and comprehensive ideas on how to improve their learning |