

CHEMISTRY 11 NAMING COMPOUNDS WORKSHEET

Write the name of the following compounds.

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| 1. AgI | 14. CCl ₄ | 27. WF ₆ |
| 2. K ₂ C ₂ O ₄ | 15. SnSO ₄ | 28. SO ₃ |
| 3. Al ₂ S ₃ | 16. HNO ₂ | 29. Ag ₂ C ₂ O ₄ |
| 4. NaOH | 17. RbI | 30. H ₃ PO ₄ |
| 5. Na ₂ SO ₃ | 18. Ca(ClO ₄) ₂ | 31. Cr ₂ O ₃ |
| 6. HNO ₃ | 19. PbCO ₃ | 32. Pb(NO ₃) ₄ |
| 7. CuSO ₄ | 20. SO ₂ | 33. Na ₂ CrO ₄ |
| 8. K ₂ Cr ₂ O ₇ | 21. KMnO ₄ | 34. NH ₄ OH |
| 9. NI ₃ | 22. H ₂ S | 35. HCl |
| 10. NH ₄ ClO ₂ | 23. Ca(OH) ₂ | 36. XeO ₂ |
| 11. HBr | 24. MgO | 37. Cu ₂ SO ₄ |
| 12. AlPO ₄ | 25. Pb(NO ₃) ₂ | 38. Pb(C ₂ O ₄) ₂ |
| 13. Cu(CH ₃ COO) ₂ | 26. NaClO ₃ | 39. Fe(ClO) ₂ |
| 40. H ₂ SO ₄ | 52. MoCl ₅ | 64. MnS ₂ |
| 41. Pb(HSO ₄) ₂ | 53. Zn(CH ₃ COO) ₂ | 65. CrO |
| 42. CoCl ₂ | 54. BaCr ₂ O ₇ | 66. (NH ₄) ₂ HPO ₄ |
| 43. Hg ₂ (NO ₃) ₂ | 55. MnO | 67. PCl ₅ |
| 44. Fe(NO ₃) ₃ | 56. K ₂ CO ₃ | 68. Sn(SO ₄) ₂ |
| 45. Na ₂ HPO ₄ | 57. Fe(OH) ₂ | 69. Cu(C ₂ H ₃ O ₂) ₂ |
| 46. SnF ₂ | 58. NaHS | 70. Hg(NO ₃) ₂ |
| 47. Li ₂ CrO ₄ | 59. H ₂ SO ₃ | 71. Al(OH) ₃ |
| 48. Sr ₃ (PO ₄) ₂ | 60. (NH ₄) ₂ SO ₄ | 72. Zn(NO ₂) ₂ |
| 49. NH ₄ HCO ₃ | 61. Al ₂ (CO ₃) ₃ | 73. Li ₂ O |
| 50. PtCl ₄ | 62. CsHCO ₃ | 74. Fe ₂ O ₃ |
| 51. CsH ₂ PO ₄ | 63. HF | 75. NaMoRe |

Answers:

1. silver iodide
2. potassium oxalate
3. aluminum sulfide
4. sodium hydroxide
5. sodium sulfite
6. nitric acid
7. copper (II) sulfate
8. potassium dichromate
9. nitrogen triiodide
10. ammonium chlorite
11. bromic acid
12. aluminum phosphate
13. copper (II) acetate
14. carbon tetrachloride
15. tin (II) sulfate
16. nitrous acid
17. rubidium iodide
18. calcium perchlorate
19. lead (II) carbonate
20. sulfur dioxide
21. potassium permanganate
22. dihydrogen sulfide
23. calcium hydroxide
24. magnesium oxide
25. lead (II) nitrate
26. sodium chlorate
27. tungsten (VI) fluoride
28. sulfur trioxide
29. silver oxalate
30. phosphoric acid
31. chromium (III) oxide
32. lead (IV) nitrate
33. sodium chromate
34. ammonium hydroxide
35. hydrochloric acid
36. xenon dioxide
37. copper (I) sulfate
38. lead (IV) oxalate
39. iron (II) hypochlorite
40. sulfuric acid
41. lead (IV) bisulfate
42. cobalt (II) chloride
43. mercury (I) nitrate
44. iron (III) nitrate
45. sodium monohydrogen phosphate
46. tin (II) fluoride
47. lithium chromate
48. strontium phosphate
49. Ammonium bicarbonate
50. Platinum (IV) chloride
51. Cesium dihydrogen phosphate
52. Molybdenum (V) chloride
53. Zinc acetate
54. Barium dichromate
55. Manganese (II) oxide
56. Potassium carbonate
57. Iron (II) hydroxide
58. Sodium hydrogen sulfide
59. Sulfurous acid
60. Ammonium sulfate
61. Aluminum carbonate
62. Cesium bicarbonate
63. Hydrofluoric acid
64. Manganese ((IV) sulfide
65. Chromium (II) oxide
66. Ammonium monohydrogen phosphate
67. Phosphorous pentachloride
68. Tin (II) sulfate
69. Copper (II) acetate
70. Mercury (II) nitrate
71. Aluminum hydroxide
72. Zinc nitrite
73. Lithium oxide
74. Iron (III) oxide
75. Nobelium molybdenum rhenium (NON-existent)