AP Physics – Photoelectric Effect

1. What energy is carried by a photon of electromagnetic radiation that has a frequency of 1.55 x 1017 Hz?
2. A photon has a wavelength of 550 nm. How much energy does this represent in Joules?
3. What is the maximum kinetic energy of a photoelectron that has been liberated from a silver metal surface by a photon that has a frequency of 3.13 x 1015 Hz?
4. What is the velocity of a photoelectron that has been liberated from a zinc metal surface by a photon that has a wavelength of 275 nm?
5. What is the cutoff wavelength for a copper metal surface?
6. 500.0 nm light is incident on a metal surface. The stopping potential is found to be 0.440 V. (a) Find the work function for this material and (b) the longest wavelength that will eject electrons from the metal.
7. CHALLENGE!: A sodium photoelectric surface with work function 2.3 eV is illuminated by electromagnetic radiation and emits electrons. The electrons travel toward a negatively charged cathode and complete the circuit shown above. The potential difference supplied by the power supply is increased, and when it reaches 4.5 V, no electrons reach the cathode.
8. For the electrons emitted from the sodium surface, calculate the following.
9. The maximum kinetic energy.

******

1. The speed at this maximum kinetic energy.

1. Calculate the wavelength of the radiation that is incident on the sodium surface.
2. Calculate the minimum frequency of light that will cause photoemission from this sodium surface.