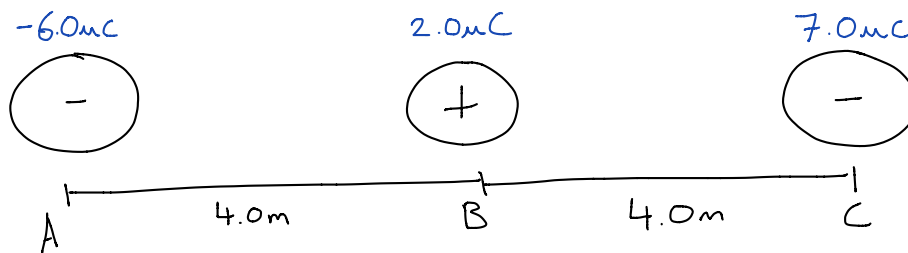


1) Electrons and protons have equal but opposite charges. The magnitude of this charge is known as the:

$$\text{Elementary Charge} = 1.60 \times 10^{-19} \text{ C}$$

A hydrogen atom contains one proton and one electron. If the electrostatic force of attraction is $8.2 \times 10^{-8} \text{ N}$, how far apart are they?

2) What is the **magnitude** and **direction** of the net force acting on the 2.0 uC charge shown below?

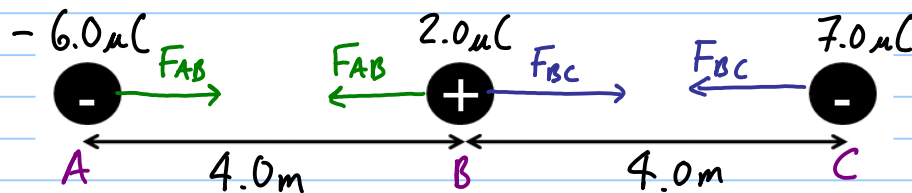


$$1.) F_E = \frac{kq_1q_2}{r^2} \quad \checkmark \quad r = \sqrt{\frac{kq_1q_2}{F_E}}$$

$$q_{p^+} = q_{e^-} = 1.60 \times 10^{-19} \text{ C} \quad r = \sqrt{\frac{(9 \times 10^9)(1.6 \times 10^{-19})(1.6 \times 10^{-19})}{8.2 \times 10^{-8}}}$$

$$= \underline{5.3 \times 10^{-11} \text{ m}} \quad \checkmark$$

2.)



$$\checkmark \quad F_{AB} = \frac{kq_Aq_B}{r^2} = \frac{(9 \times 10^9)(6.0 \times 10^{-6})(2.0 \times 10^{-6})}{(4.0)^2} = 6.75 \times 10^{-3} \text{ N}$$

$$F_{BC} = \frac{kq_Bq_C}{r^2} = \frac{(9 \times 10^9)(2.0 \times 10^{-6})(7.0 \times 10^{-6})}{(4.0)^2} = 7.875 \times 10^{-3} \text{ N}$$

$$F_{\text{net}} = F_{BC} - F_{AB} = 7.875 \times 10^{-3} - 6.75 \times 10^{-3}$$

$$= \underline{1.1 \times 10^{-3} \text{ N (right)}} \quad \checkmark$$